LANKOVITS, A.V.

Determination of the size of the fetus in labor and some factors in its development. Vop. okh. mat. i det. 6 no.10:44-49 0 '61.

1. Iz kafedry akusherstva i ginekologii (zav. - chlen-korrespondent AMN SSSR prof. L.S.Persianinov) II Moskovskogo meditsinskogo instituta imeni N.I.Pirogova.

(FETUS)

LANKOVITS, A.V.

Some terms used in obstetrics. Vop. okh. mat. i det. 7
no.1:70-74 Ja '62.

1. 1z kafedry akusherstva i ginekologii lechebnogo fakul'teta
(zav. - chlen-korrespondent AMI SSSR prof. L.S. Persiaminov)
II Moskovskogo meditsinskogo instituta imeni N.I. Pirogova.

(OBSTETRICS--TERMINOLOGY)

MATSPANOVA, O.D., kand. med. nauk; LANKOVITS, A.V., prof.; KRASOVSKIY, Ye.B., doktor med. nauk, red.; LIBENZON, L.L., kand. med.nauk, red.

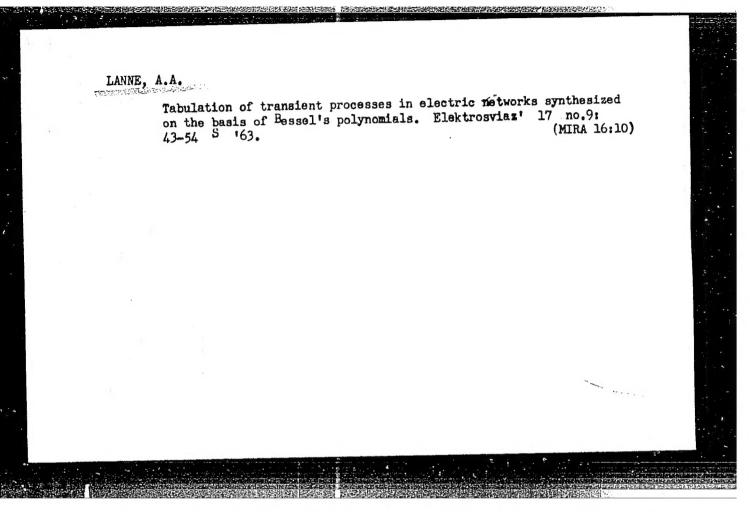
[Authors abstracts of scientific papers completed in 1961] Avtoreferaty nauchnykh rabot, vypolnennykh v 1961 g. Red. koll.: O.D.Matspanova i dr. Moskva, 1962. 118 p. (MIRA 16:11)

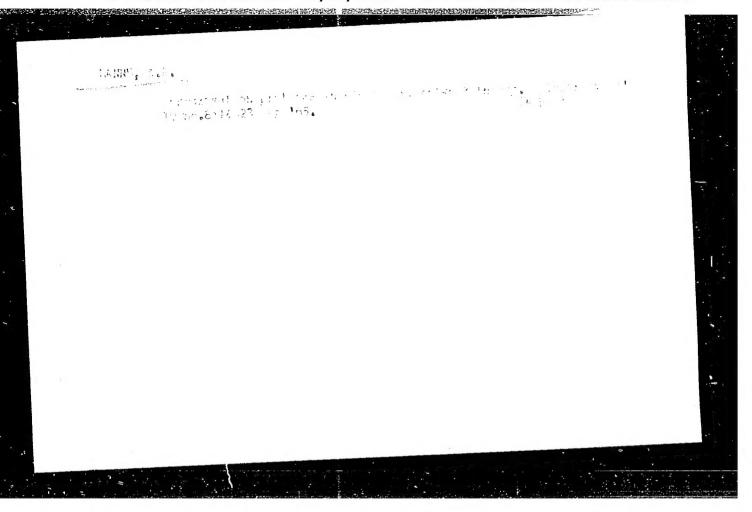
l. Moscow. (Province) Oblastnoy nauchno-issledovatel'skiy institut akusherstva i ginekologii. 2. Direktor Moskovskogo oblastnogo nauchno-issledovatel'skogo instituta akusherstva i ginekologii (for Matspanova). 3. Zamestitel' direktora po nauchnoy chasti Moskovskogo oblastnogo nauchno-issledovatel'skogo instituta akusherstva i ginekologii (for Lankovits). (OBSTETRICS) (GYNECOLOGY) (PEDIATRICS)

LANKOVITS, A.V.

Gesarean sections in Moscow Province in 1961. Vop. okh. mat. i det. 8 no.7:57-63 Jl '63. (MIRA 17:2)

l. Iz Moskovskogo oblastnogo nauchno-issledovatel'skogo instituta akusherstva i ginekologii (direktor - kand. med. nauk O.D. Matspanova, nauchnyy rukovoditel' - prof. A.V. Lankevits).

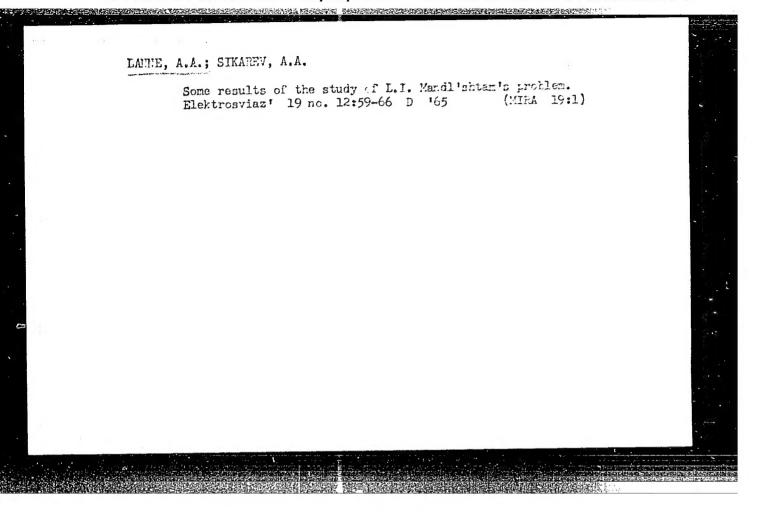




ARZHANNIKOV, Ye.P.; LAKNI, A.A.

Optimal characteristics of low-frequency filters. Rediotekhnika 20 no.10:21-30 0 65. (MFM 18:11)

1. Deystvitel'nyye ohleny Nauchno-tekhnicheskogo obshchestva radiotekhniki i elektrosvyazi.



L 47212-66

ACC NR. AR6019063

SOURCE CODE: UR/0274/66/000/001/A008/A008

AUTHOR: Lanne, A. A.; Okunev, Yu. B.; Sikarev, A. A.

REF SOURCE: Tr. uchebn. in-tov svyazi. M-vo svyazi SSSR, vvp. 24, 1965, 49-58

TITLE: Statistical evaluation of one class of phase-keyed signals

SOURCE: Ref. zh. Radiotekhnika i elektrosvyaz', Abs. 1A46

TOPIC TAGS: signal analysis, phase shift

TRANSLATION: A group signal in a phase-keyed channel can be represented by the follow-

ing equation: $z(t) = \sum_{k=k_1}^{k_1} A_k \cos(k\omega_0 t + \varphi_k + \Delta \varphi_k),$

where $\omega_0 = \frac{2\pi}{T}$; $k_2 - k_1 + 1 = n$ is the number of the phase components, T is the pulse width, ϕ_k is the initial phase of the k-th component, and $\Delta\phi_k$ is the random discrete phase shift of the k-th signal component. In the case of determined signals, the peak fact-

or is expressed by

Card 1/2

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UDC: 621.391.133

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ACC NR: AR6019063	63
where $S_0 = \min_{\substack{q_k \ 0 < l < T}} \max_{\substack{k=k_1 \ k=k_1}} \sum_{k=k_1}^{k_2} A_k \cos(k\omega_0 l + q_k)$	
and $S_0 < C \sqrt{\ln n} \sqrt{\sum_{k=k_1}^{k_1} A_k^2};$	
\mathcal{C} is an absolute constant. It was established that for a group phase-keyed signal with a number of components $n>6$, the selection of the original phases do not affect the statistical properties of the signal. Evaluation of statistical signal characteristics are also presented and examples illustrating the application of the results are given. 5 figures, 1 table, 1 reference. N. G.	
SUB CODE: 17/ SUBN-DATE: none	
 Card 2/2 ^{fv}	

Morphology of the gynoecium and the fruit of geranium.

Nauch. dokl. vys. shkoly; biol. nauki no.4:104-109

'63. (MIRA 16:11)

1. Rekomendovana kafedroy vysshikh rasteniy Moskovskogo gosudarstvennogo universiteta im. Lomonosova.

MUSHKALO, L.K.; SHEYKO, D.1.; LANOVAYA, Ye.I.

Condensat' of c-aminocelenophonol with unsaturated ketones.
Report No 2. Ukr.khim.zhur. 30 no.5:502-503 '64.

1. Kiyevskiy gosudarstvennyy universitet.

(MIRA 18:4)

MUSHKALO, L.K.; LANOVAYA, Z.I.

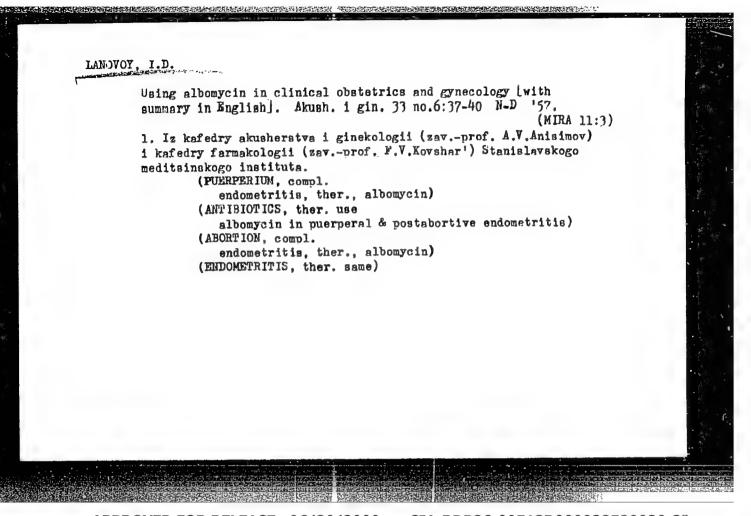
Condensation of unsaturated carbonyl compounds and β-halo ketones with β-aminoethylmercaptans. Ukr.khim.zhur. 21 no.5:631-635 '55.

(MIRA 9:3)

1. Kiyevskiy gosudarstvennyy universitet imeni T.G. Shevchenko, Kafedra organicheskoy khimii.

(Carbonyl compounds) (Ketones) (Thiols)

LANOVOY, I. D.: Master Med Sci (diss) -- "The use of the Soviet antibictic albomycin in treating endometritis (Experimental-clinical investigation)". Kiev, 1956. 20 pp (Kiev Order of Labor Red Banner Med Inst im Acad A. A. Bogomolets), 200 copies (KL, No 7, 1959, 129)



IVANOVA, T.I.; LANOVCY, I.D.; ASMOLOVSKIY, G.V.; FEDOROV, R.V.

Therapeutic effect of monomycin in experimental endometritis.

Antibiotiki 9 no.5:462-463 My '64. (MIRA 18:2)

1. Iva o-Frankovskiy meditsinskiy institut.

IVANOVA, T.I., prof.; VIKTOROVSKAYA, Ye.N., dotsent; LAKOVOY, I.D.;
KRIVOSHEYEVA, M.V.

Use of albomycin in treating women with inflammatory diseases of the genitalia. Sov.med. no.3:121-122 '62. (MIRA 15:5)

1. Iz kafedry akusherstva i ginekologii (zav. - prof. A.V.
Anisimov) i kafedry mikrobiologii (zav. - prof. T.I. Ivanova)
Stanislavskogo meditsinskogo instituta (dir. - dotsent G.A.
Babenko).

(GENERATIVE ORGANS, FEMALE—DISEASES)

(ALBOMYCIN)

S/190/60/002/009/012/019 B004/B060

AUTHORS:

Lanovskaya, L. M., Gantmakher, A. R., Medvedev, S. S.

TITLE:

Polymerization of Ethylene by Means of the Combined Catalyst α-TiCl₃ - AlR₃ in the Presence of Various Monomers.

I. The Effect of Various Monomers on the Polymerization of

Ethylene

PERIODICAL:

Vysokomolekulyarnyye soyedineniya, 1960, Vol. 2, No. 9,

pp. 1391-1397

TEXT: The authors wanted to study the interaction of various unsaturated compounds with the combined catalyst, and its effect on the polymerization of ethylene under conditions at which these compounds still polymerize at a negligibly low rate. The authors describe the purification of the reagents, the reaction vessel (Fig. 2) with magnetic stirrer and

Card 1/4

Polymerization of Ethylene by Means of the Combined Catalyst $\alpha\text{-TiCl}_3$ - AlR $_3$ in the Presence of Various Monomers. I. The Effect of Various Monomers on the Polymerization

S/190/60/002/009/012/019 B004/B060

thermostat, and a device (Fig. 1) which served for introducing the octane solvent and the $(i-c_4H_9)_3$ Al into the reaction vessel. The measure-

ments were made at a constant ethylene pressure of 200 torr by the method developed by A. I. Gel'bshteyn and M. I. Temkin (Ref. 8). The experimental procedure was worked out by Gritsenko and Lanovskaya. α-methyl styrene, isoprene, butadiene, and isobutylene were used as admixtures. In the first series of experiments (Table 1, Fig. 3), the monomer was filled into the reaction vessel before introducing the ethylene. In the second series of experiments (Tables 1,2, Figs. 4-6), the ethylene was first polymerized during two hours, the monomer was then added, and polymerization was carried on for five more hours. In the experiments specified in Table 1, the authors used TiCl₃ which was ob-

tained from TiCl by reduction by means of antimony. Table 2 specifies

Card 2/4

of Ethylene

Polymerization of Ethylene by Means of the Combined Catalyst $\alpha\text{-TiCl}_3$ - AlR_3 in the Presence of Various Monomers. I. The Effect of Various Monomers on the Polymerization of Ethylene

S/190/60/002/009/012/019 B004/B060

the experiments in which TiCl₃ was produced by the reduction of TiCl₄ by means of titanium metal. Experiments revealed that the polymerization rate of ethylene is retarded in the presence of one of the monomer compounds mentioned. The molecular weight of the resulting polyethylene is, however, not influenced thereby. As to their reaction-retarding effect, the various monomer compounds are mentioned in the order butadiene, isoprene > styrene > isobutylene > α -methyl styrene. Diene hydrocarbons, thus, have the greatest retarding effect. The addition of monomers prior to or after the beginning of polymerization bears no influence on this effect. The authors mention a discussion by A. R. Gantmakher on a lecture by A. A. Korotkov at the International Symposium in Prague, 1957. There are 6 figures, 2 tables, and 8 references: 2 Soviet, 4 US, and 2 German.

Card 3/4

Polymerization of Ethylene by Means of the Combined Catalyst α -TiCl₃ - AlR₃ in the

S/190/60/002/009/012/019 B004/B060

Presence of Various Monomers, I. The Effect of Various Monomers on the Polymerization of Ethylene

ASSOCIATION:

Fiziko-khimicheskiy institut im. L. Ya. Karpova

(Physico-chemical Institute imeni L. Ya. Karpov)

SUBMITTED:

April 11, 1960

Card 4/4

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15.8101

S/190/60/c02/011/010/027 B004/B060

AUTHORS:

Lanovskaya, L. M., Gantmakher, A. R., Medvedev, S. S.

TITLE:

Polymerization of Ethylene by Means of Combined α-TiCl3-AlR3

Catalyst in the Presence of Various Monomers II Some Problems Concerning the Polymerization Mechanism in the

Presence of Combined Catalysts

PERIODICAL:

Vysokomolekulyarnyye soyedineniya, 1960, Vol. 2, No. 11.

pp. 1655 - 1658

TEXT: This is a discussion of the results obtained by the authors in Ref. 1 concerning the effect of various monomers on the polymerization of ethylene by α -TiCl₃-AlR₃ catalysts. The authors' experiments revealed that additions

of isobutylene, styrene, isoprene, or butadiene reduce the polymerization rate of ethylene, complexes of these monomers being formed on the catalyst surface. The ability to form complexes is reduced in the series butadiene isoprene styrene isobutylene a-methyl styrene. This succession is analogous to the series obtained by other researchers for Card 1/3

85414

Polymerization of Ethylene by Means of S/190/60/002/011/010/027 Combined α-TiCl₃-AlR₃ Catalyst in the Presence B004/B060 of Various Monomers. II. Some Problems Concerning the Polymerization Mechanism in the Presence of Combined Catalysts

compounds of platinum, silver, and other metals. The authors base on their experimental results to conclude that the monomers react with the titanium component of the catalyst. A reaction with the aluminum component, which is a Lewis acid, would yield another series of activities. The following reaction scheme is given:

 $TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{1} \\ R & k_{2} \end{pmatrix} M \cdot TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_{3} \\ M_{n}R & k_{3} \end{pmatrix} TiCl_{3} \cdot Al = \begin{pmatrix} R & k_$

Polymerization by combined catalysts thus does not have a typical anionic course, but is a more complicated process. This has some resemblance with polymerization in the presence of lithium alkyls, but differs from it by specific properties which depend on the structure of the combined catalyst. The authors mention A. A. Babushkin, L. A. Gribov, and A. D. Geliman. There are 14 references: 5 Soviet, 4 US, 3 British, 1 French, and 1 German.

Card 2/3

85414

Polymerization of Ethylene by Means of \$/190/60/002/011/0:0/027 Combined a-TiCl_3-AlR_ Catalyst in the Presence B004/B060 of Various Monomers, II. Some Problems Concerning the Polymers

of Various Monomers. II. Some Problems Concerning the Polymerization Mechanism in the Presence of Combined Catalysts

ASSOCIATION: Fiziko-khimicheskiy institut im. L. Ya Karpova (Physico-chemical Institute imeni L. Ya Karpov)

SUBMITTED: May 5, 1960

Card 3/3

LANOVSKAYA, L.M.; MAKLETSOVA, N.V. [deceased]; GANTMAKHER, A.R.;

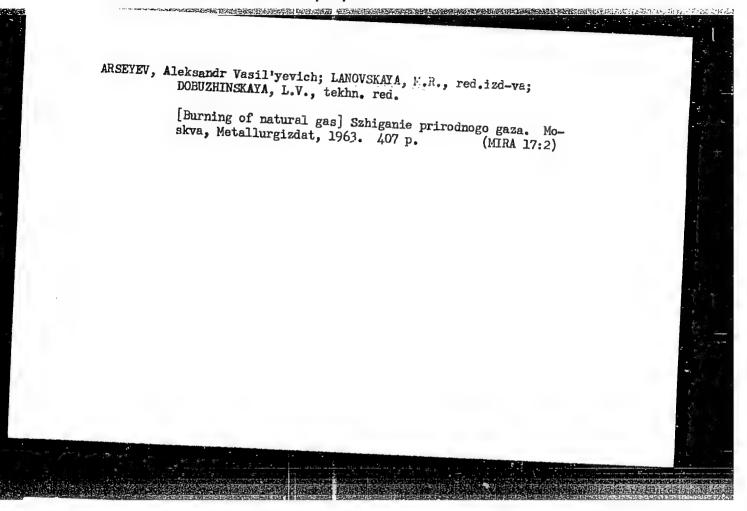
MEDVEDEV, S.S.

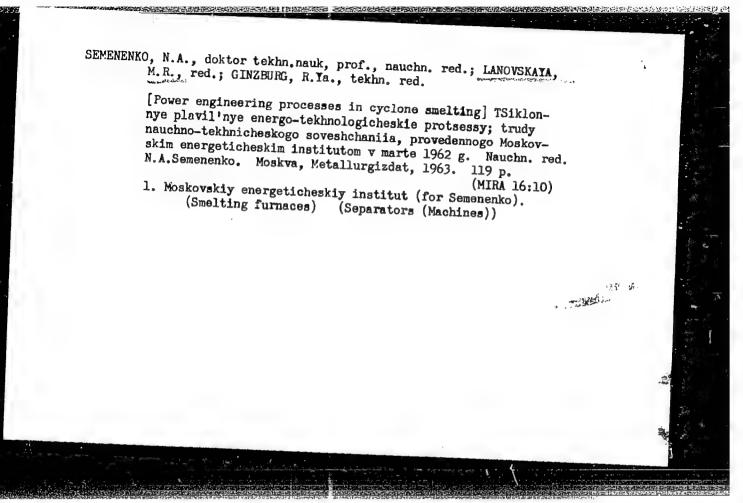
Polymerization of ethylene in the presence of various composite catalysts based on TiCl₃. Vysokom. soed. 7 no.4:741-746 Ap 165.

Nature of the active centers in the processes of polymerization in the presence of composite catalysts based on TiCl₃. Ibid.: 747-750

(MIRA 18:6)

1. Fiziko-khimicheskiy institut imeni Karpova, Moskva.

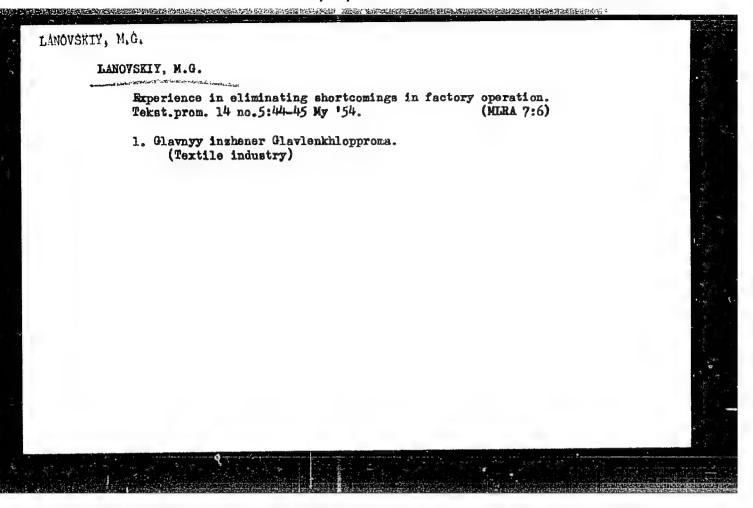


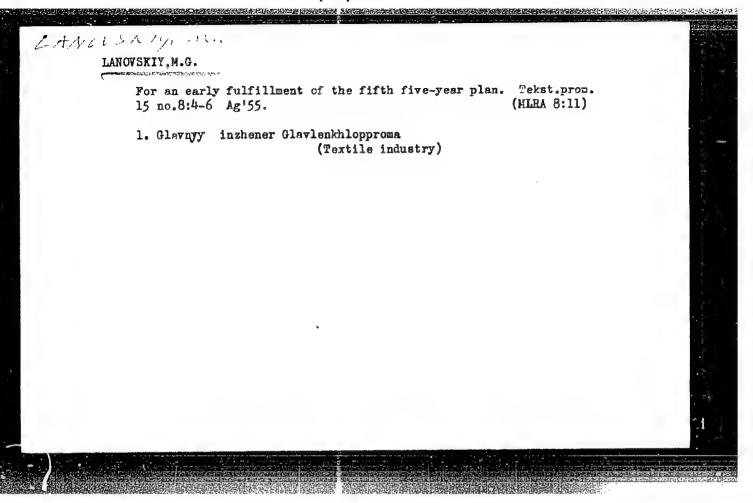


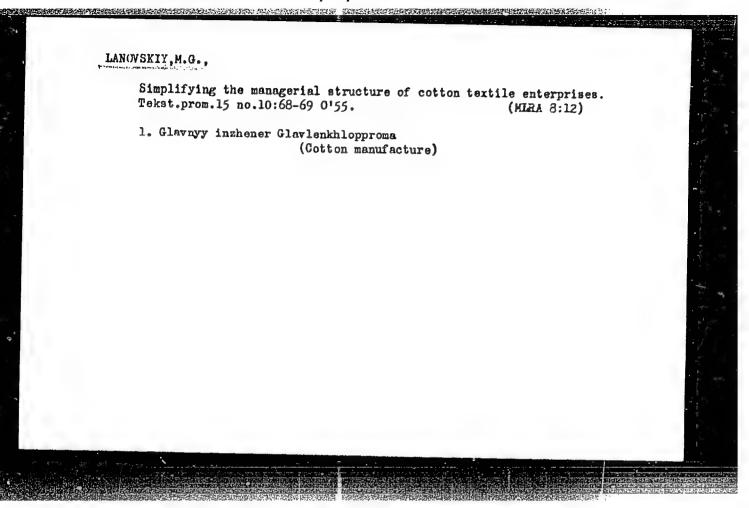
NEPOMNYASHCHIY, Igor' Lazarevich; BURSHTEYN, M.D., red.; LANOVSKAYA,
M.R., red. izd-va; ATTOPOVICH, M.K., tekhn. red.[deceased]

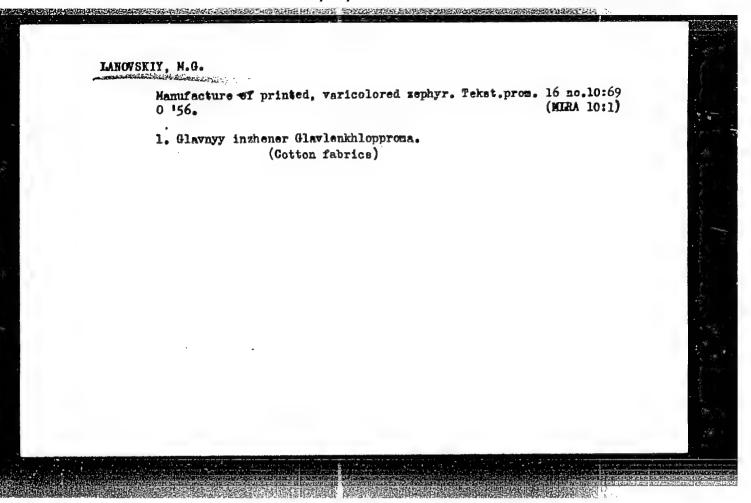
[Design and construction of coking machines]Koksovye mashiny,
ikh konstruktsii i raschety. Izd.2., perer. i dop. Moskva,
Metallurgizdat, 1963. 388 p. (MIRA 16:2)

(Coking plants—Equipment and supplies)









LANOVSKIY, M. G.

Development of the textile industry in the Leningrad Economic Region. Tekst.prom.17 no.11:16-19 N '57. (MIRA 10:12)

1. Glavnyy inzhener Upravleniya tekstil'noy promyshlennosti Leningradekogo sovnarkhoza.

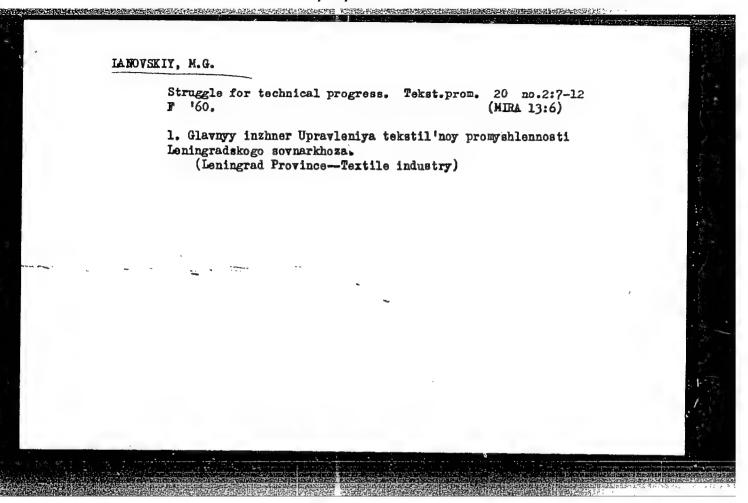
(Leningrad economic region--Textile industry)

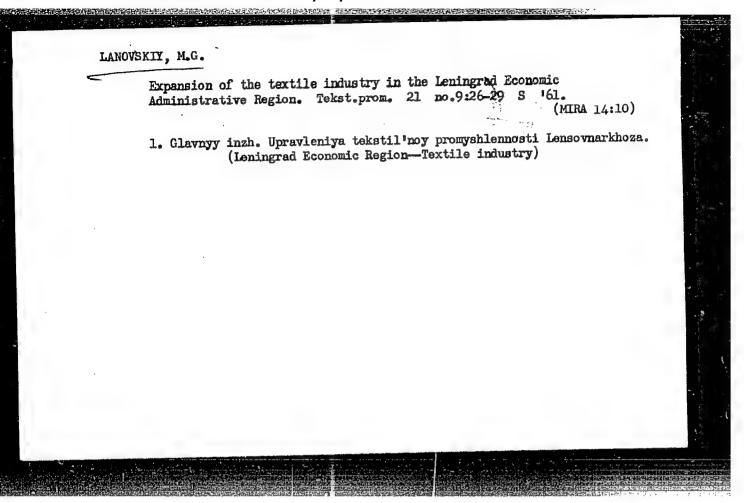
LANOVSKIY, M.G.

Improving the assortment and increasing the output of fabrics for children's clothing. Tekst. prom. 18 no.8:17-20 Ag '58.

(MIRA 11:10)

1.Glavnyy inzhener upravleniya tekstil'noy promyshlennosti Leningradskogo sovnarkhoza. (Textile fabrics) (Children's clothing)



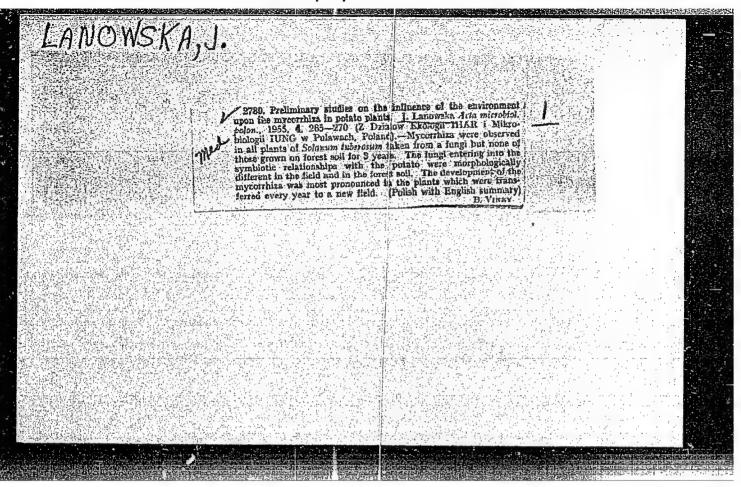


IANOVSKIY, M.G. On the road toward complete automation of production processes. Tekst.prom. 22 no.6:5-8 Je '62. (MIRA 16:5) 1. Glavnyy inah. Upravleniya tekstil'noy promyshlennosti Leningradskogo soveta narodnogo khozyaystva. (Textile industry) (Automation)

LANOVSKIY, M.G., red.; SKOL'NIK, I.D., red.

[Scientific and technial contest papers of the members of the Scientific and Technical Society of the Textile Industry for the period from 1962 to 1963; materials on an exchange of experience in production technology) Konkurenye nauchnotekhnicheskie raboty chlenov NTO tekstil'noi promyshlennosti za 1962 - 1963 g.; materialy po obmemu proizvodstvennotekhnicheskim opytom. Leningrad, Neuchnotekhnich ob-vo legkoi promyshl. Leningraphana, 1964. 89 p.

(MIRA 18:4)



LANOWSKA, Jadwiga

Investigations concerning the appearance of mycorrhiza in potatoes of the Lorch variety in field and forest clearing plantations. Rocz nauk roln rosl 82 no.3:779-804 161.

1. Zaklad Ekologii Rolniczej, Instytut Uprawy, Nawozenia i Gleboznawstwa, Pulawy.

KARPIRKO, B.K., kend, tekhn. nauk, IVANOVA, I.G., inch.; IANOVOY, V.G., inch.; SHOHFRHIMA, B.A., inch.

A d.c. motor with printed armature winding. huerg. i elektricals. prom. no.3:33-36 JI-C '65. (MIRA 19:9)

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IANSHCHIKOV, M.T.; LAPTEV, V.L., starshiy inzh.

Increase in the protection of automatic block systems of electrified

Increase in the protection of automatic block systems of electrified railroad districts against overvoltages caused by lightning strokes.

Avtom., telem.i sviaz: 6 no.4:32-33 Ap :62. (MJRA 15:4)

1. Nachal'nik laboratorii signalizatsii i svyazi Sverdlovskoy dorogi, vneshtatnyy korrespondent zhurnala "Avtomatika, telemekhanika i svyazi" (for Lanshchikov). 2. Laboratoriya signalizatsii i svyazi Sverdlovskoy dorogi (for Laptev). (Railroads—Signaling—Block system) (Electric protection)

LANSECHIKOV, M.T.; ALYAKIN, G.A.

A device for locating electric lines. Avtom., telem. i sviaz' 6 no.7:
(MIRA 16:2)

1. Nachal*nik laboratorii signalizatsii i svyazi Sverdlovakoy dorogi, vneshtatnyy korrespondent zhurnala "Avtomatika, telemekhanika i svyaz'" (for lanshchikov). 2. Starshiy inzh. laboratorii signalizatsii i svyazi Sverdlovskoy dorogi (for Alyakin).

(Electric lines-Underground) (Electric lines-Measurement)

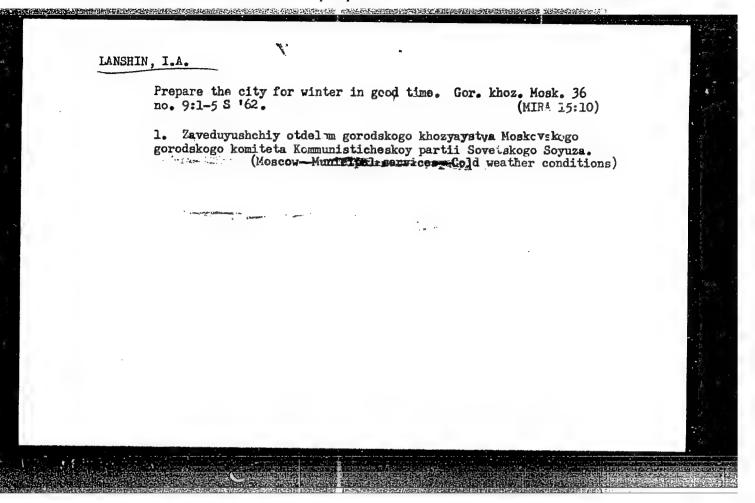
Lan'shin, A.P.

Activate the work of production conferences. Put' i put, khoz. no.3:1-2
Mr '53. (Mira 11:4)

1. Sektetar' TSentral'nogo komiteta profsoyuza rabochikh
zheleznodorozhnogo transnorta.

(Railroads)

Municipal economy of the capital. Gor.khoz.Mosk. 35 no.9:15-20 S '61. (MIFA 14:10) 1. Zaveduyushchiy otdelom Moskovskogo gorodskogo komiteta Kommunisticheskoy partii Sovetskogo Soyuza. (Moscow—Municipal services)



SHAKHPARCHOV, M. I. and LANSHINA. L. V.

"The Microstructure of Sound,"

report presented the 6th Sci. Conference on the Application of Ultrasound in the Investigation of Matter, 3-7 Feb 58, Moscow, organized by Min. of Education REFER and Moscow Oblast Pedagogic Inst. im N. K. Krupskaya

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S/020/60/133/003/029/C31/XX B004/B064

AUTHORS:

Lanshina, L. V., and Shakhparonov, M. I.

TITLE:

The Fine Structure of the Rayleigh Dispersion of Light in Solutions and the Dispersion of Hyperacoustic Vibrations

PERIODICAL:

Doklady Akademii nauk SSSR, 1960, Vol. 133, No. 3, pp. 624

TEXT: The apparatus used by the authors to photograph the fine structure of the Rayleigh line of dispersed light has already been described in Ref. 1. The authors investigated mixtures of acetone and water at 25°C with a molar ratio of acetone of 0.0; 0.06; 0.2; 0.4; 0.7, or 1.0, and mixtures of water and methanol with a molar ratio of the latter amounting to 0.15; 0.36; 0.6; or 1.0. Measurements were made by means of the Hg line λ = 4358 A, and for pure water also with λ = 4046 A. The photographs were photometrically treated with an $M\dot{\phi}$ -4 (MF-4) microphotometer. Fig. 1 shows the propagation velocity of hyperacoustic vibrations in acetone - water ($\omega \approx 0.6 \cdot 10^{10}$ sec-1, $\Lambda \approx 22 \cdot 10^{-6}$ cm) and methanol - water mixtures ($\omega \approx 0.5 \cdot 10^{10}$ sec⁻¹,

Card 1/3

85524

The Fine Structure of the Rayleigh Dispersion of Light in Solutions and the Dispersion of Hyperacoustic Vibrations

S/020/60/133/003/029/031/XX B004/B064

 $\Lambda \approx 22.0 \cdot 10^{-6}$ cm) as computed from the equation $\Delta V/V = 2n(v/c)\sin(\theta/2)$. The data of Refs. 4, 5 on the propagation velocity of ultra-acoustic vibrations ($\omega \approx 56 \cdot 10^{-5}$ sec⁻¹, $\Lambda \approx 24 \cdot 10^{-3}$ cm) are added for comparison. A considerable negative dispersion of hyperacoustic vibrations was found to exist; $\Delta V/V$ reached about 5%, while the root mean square of deviation was only about 1.4%. The negative dispersion vanishes with falling water concentration x_1 ; i.e., for acetone – water at $x_1 = 0.65$ and for methanol – water at $x_1 = 0.5$. This negative dispersion is explained by the structural relaxation of water, and corresponds to the well-known anomalies of water. Moreover, the integral intensity I_c of the central component and the intensity I_t of the translational component by Mandel shtam and Brillouin were compared, and the refractive index n_λ was represented as a function of concentration (Fig. 2). Considerable positive deviations from the Raoult law were observed in the acetone – water mixture. The fluctuations of concentration attain their

Card 2/3

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The Fine Structure of the Rayleigh Dispersion S/020/60/133/003/029/031/XX of Light in Solutions and the Dispersion of B004/B064

Hyperacoustic Vibrations

maximum at $x_1 \approx 0.4$. In this range of concentration, n_{λ} passes through a maximum and $I_c/2I_{tr}$ through a minimum. In the methanol - water mixture which is almost ideal, the fluctuations of concentration are slight. $I_c/2I_{tr}$ passes through a maximum in the same range of concentration in which n_{λ} attains its maximum. There are 2 figures and 6 references: 5 Soviet and 1 Indian.

ASSOCIATION: Moskovskiy gosudarstvennyy universitet im. M. V. Lomonosova

(Moscow State University imeni M. V. Lomonosov)

PRESENTED: March 7, 1960 by V. V. Shuleykin, Academician

SUBMITTED: March 7, 1960

Card 3/3

LANSHINA L.V.

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PHASE I BOOK EXPLOITATION SOV/5469

Soveshchaniye po kriticheskim yavlenida i flyuktuatsiyam v rastvorakh. Moscow, 1960.

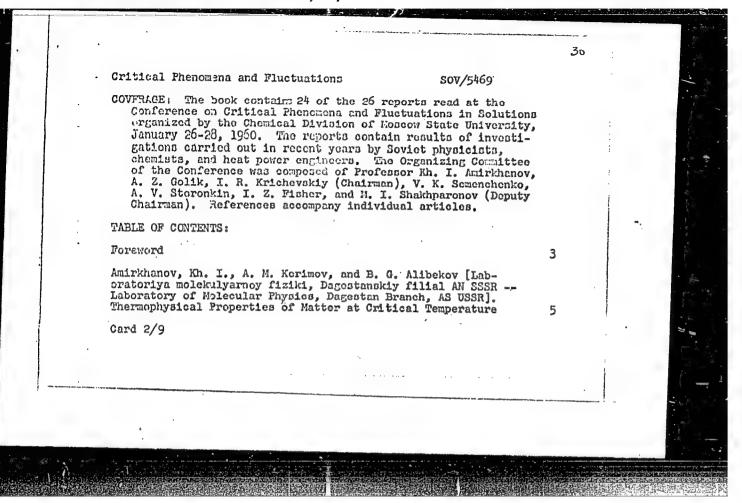
Kriticheskiye yavleniya i flyuktuatsii v rastvorakh; trudy sovesicheniya, yanvar' 1960 g. (Critical Phonomena and Fluctuations in Solutions; Transactions of the Conference, January 1960) Moscow, Izd-vo AM SSSR, 1960. 190 p. 2,500 copies printed.

Sponsoring Agencies: Akademiya nauk SSSR. Otdeleniyo khimicheskukh nauk. Koskovskiy gosudarstvennyy universitet im. M. V. Lomonosova. Khimicheskiy fakul'tet.

Responsible Ed.: M. I. Shakhparonov, Doctor of Chemical Sciences, Professor; Ed. of Publishing House: E. S. Dregunov; Tech. Ed.: S. G. Tikhomirova.

PURPOSE : This collection of articles is intended for scientific personnel concerned with chemistry, physics, and heat power engineering.

Card 1/9



Zatscpina, L. P., and H. I. Shakharonov [Laboratory of the Physical Chemistry of Solutions, Garristry Division, Hoscow State University imeni H. V. Lomonosov]. Rayleigh Hight Scattering in Nitrobenzene Cyclohexane and Ethyl Alchol Diethylamine Solutions Kasimov, R. M., and M. I. Shakharonov [Laboratory of the Physical Chemistry of Solutions, Chemistry Division, Hoscow State University imeni M. V. Lomonosov]. Dielectric Properties of Solutions in Electromymetic Fields of the Millimetric Band and Concentration Fluctuations Krichevskiy, I. R., and N. Ye. Khazanova [Laboratoriya vysokikh davleniy, GIAP Laboratory of High-Pressure [Studies], Hoscow Stati Design and Planning Scientific Research Institute of the Nitrogen Industry]. Diffusion of Liquid and Gaseous Solutions in the Critical Region Krichevskiy, I. R., and Yu. V. Tsekhanskaya [Laboratory of Card 4/9	Gritical Phenomena and many	30	
	Zatsepina, L. P., and II. I. Shakhparonov [Laboratory of the Physical Chemistry of Solutions, Cachistry Division, Moscow State University imeni M. V. Lomonosov]. Rayleigh Light Scattering in Nitrobenzene Cyclohexane and Ethyl Alchol Diethylamine Solutions Kasimov, R. H., and M. I. Shakhparonov [Laboratory of the Physical Chemistry of Solutions, Chemistry Division, Moscow State University imeni M. V. Lomonosov]. Dielectric Properties of Solutions in Electromagnetic Fields of the Millimetri Band and Concentration Fluctuations Krichevskiy, I. R., and N. Ye. Khazanova [Laboratoriya vysok davleniy GIAP Laboratory of High-Pressure [Studies], Moscow State Design and Planning Scientific Research Institute of the Nitrogen Industry]. Diffusion of Liquid and Gaseous Solutions in the Critical Region Krichevskiy, I. R., and Yu. V. Tsekhanskaya [Laboratory of	32 ikh of	-
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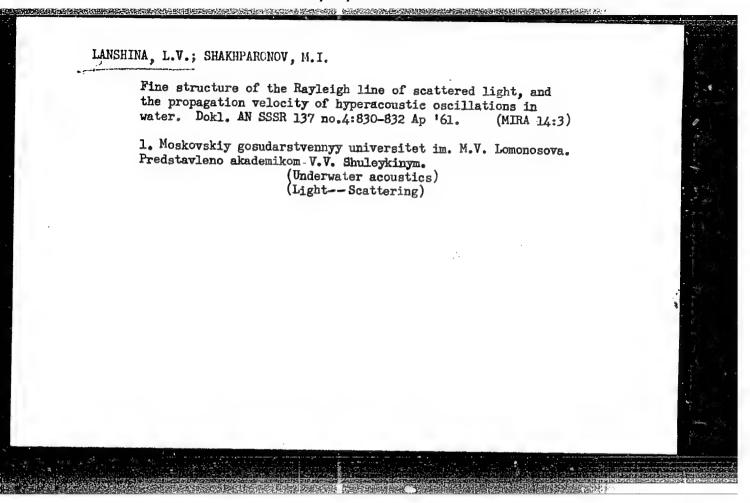
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Critical Ph	enomena and Fluctuations	Sov/5	469		
High-Prossu Processes in	re [Studies], GIAP]. Kinetics on the Critical Region	f Heterogeneous	54		
oratory of	, I. R., N. Ye. Khazarava, and L. High Prossure [Studies], GIAP]. in the Critical Region of Liqui	Liquid-Vapor	61	:	
Physical Che State Univer	N, and <u>M. I. Shakhnaronov</u> [Labor emistry of Solutions, Chemistry rsity imeni M. V. Lemonosov]. Ptructure of Solutions	Division, Moscow	73		-
Physical Che State Univer	. V., and M. I. Shakhparonov [Lo emistry of Solutions, Chemistry reity imeni M. V. Lemenosov]. T Rayleigh Light Scattering in So	Division, Moscow hin Structure of	77		
	V., and Ya. M. Labkovskiy (Kafed ropetrovskiy gosudarstvomyy univ				
Card 5/9				٠.	
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	Critical Phenomena and Fluctuations Sov/5469				
	ment of Experimental Physics, Dacpropetrovsk State University]. Investigation of Density Fluctuations in Ether and Benzeno Rased on X-Ray Scattering at Marrow Angles	81	· ·		
i .	Mokhov, N. V., and I. V. Kirsh [Department of Experimental Physics, Enepropetrovsk State University] Variation in the Sizes of Concentration Fluctuations in Relationship to Temperature and Concentration in Binary Liquid Systems Having an Upper Critical Dissolving Temperature	89			57 - 6 20 - A 11 - A 12
	Nozdrev, V. F., B. I. Kallyanov and H. G. Shirkevich [Noskov-skiy oblastnoy pedagogichecky institut Pedagogical Institute of the Mczeow Oblast]. Hypersonic Investigation in Organic Liquids at Constant Density in the Vicinity of the Critical State	93		Φ.,	The second second
	Rott, L. A. [Minskiy lesotekhnicheskiy institut Minsk Forestry Engineering Institute]. Concerning the Diffusion in the Critical Stratification Region	102			The said and a
	Card 6/9		1		
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		30	
	Critical Phenomena and Fluctuations SOV/5469		
	Roshchina, G. P. [Laboratoriya molekulyarnoy fiziki, Fizi- cheskiy fakul'tet, Kiyevskiy gosudarstvennyy universitet im. T. G. Shevchenko Laboratory of Molecular Physics, Divi- sion of Physics, Kiyev State University imeni T. G. Shevchenko] Investigation of Fluctuations in Solutions by the Method of Light Scattering	109	
	Skripov, V. P. [Laboratoriya molekulyarnoy fiziki, Uraliskiy rolitekinicheskiy institut im. S. M. Kirova Laboratory of Rolecular Physics, Ural Polytechnic Institute imeni S. M. Molecular Physics, Ural Pelytechnic Institute imeni S. M. Kirov]. Special Structural Features of Matter in the Vicin-Kirov]. Special Structural Features of Phenomena ity of the Critical Point and Transfer Phenomena	117	
	Skripov, V. P., and Yu. D. Kolpakov [Laboratory of Molecular Physics, Ural Polytechnic Institute imeni S. M. Kirov, and Physics, Ural Polytechnic Institute imeni S. M. Kirov, and the Laboratoriya teplofiziki, Ural skiy filial in SSSR Thermophysics Laboratory, Ural Branch, AS USSR]. Light Scattering in Carbon Dioxide along Pre- and Post-Critical Isotherms Smirnov, B. A. [Institut neftekhimicheskogo sinteza AN SSSR	126	
·	Card 7/9	;	
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Critical Phenomena and Fluctuations SOV/5469			
Institute of Petrochemical Synthesis, AS USSR (Moscow) Visua Observations in the Critical Region	1 137	: (1)	
Fisher, I. Z., and V. K. Prokhorenko. Concerning the Fluct- uations of Coordination Numbers in Liquids	142		
Ficher, I. Z. [Belorusaldy Cosudarstvennyy Universitet Belorussian State University (Minsk)] Correlation Analysis of the Critical Point	148		
Shakhparonov, M.I. [Laboratory of the Hydical Chemistry of Solutions, Chemistry Division, Moscow State University imeni M. V. Lomonosov]. Fluctuations in Solutions	151		
Shimanskaya, Ye. T., and A. Z. Golik [Laboratory of Molecular Physics, Physics Division, Kiyev State University imeni T. G. Shevchenko]. Investigation of the Critical State, Liquid-Vapor, of Solutions by Tepler's Method	161		
Card 8/9			
		ns and the supplementary	
Card 8/9			

	Critical Phenomena and Fluctuations SOV/51:69			•
	Shimanskaya, Ye. T., Yu. I. Shimanskiy, and A. Z. Golik [Latoratory of Molecular Physics, Division of Physics, Kiyev State University imeni T. G. Shevchenko]. Investigation of the Critical State of Pure Substances by Tepler's Method	171		
	Resolution of the Conference on Critical Phenomena and Fluctuetions in Solutions	- 189	,	
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SHAKHPARONOV, M.I.; TUNIN, M.S.; LANSHINA, L.V.; SUKHOTINA, G.G.

Hyperacoustic properties of liquids and molecular structure.

Ukr.fiz.zhur. 7 no.7:792-796 Jl '62. (MIRA 15:12)

1. Moskovskiy universitet.

(Sound—Speed) (Molecules)

APPROVED FOR RELEASE: 06/20/2000 CIA-RDP86-00513R000928520020-3"

S/185/62/007/007/010/010 I048/1248

AUTHORS:

Shakhparonov, M.I., Tunin, M.S., Lanshina, L.V., and

Sikhotina, G.G.

TITLE:

The hyperacoustic properties of liquids and the

structure of molecules

PERIODICAL:

Ukrains'kyy fizychnyy zhurnal, v.7, no.7,

1962, 792-796

TEXT: The dispersion of sound velocities in the hyperacoustic range was studied in a number of pure liquids, using the technique described by I.L. Fabelinskiy in UNF 63, 355, 1957. The experiments were carried out at 20-85°C and were based on the examination of the fine structure of the 4538 A Rayleigh line. The

Card 1/3

S/185/62/007/007/010/010 I048/I248

The hyperacoustic properties of ...

absorption of ultrasonic waves with frequencies of 8.5 - 34.4 Mc/scc was also measured. Accuracy was ±0.3 - 1% in the ultrasonic, and ±2 - 3% in the hypersonic ranges. Dispersion of the sound velocities was observed in the following media: carbon disulfide, methylene chloride, carbon tetrachloride, thiophene, furan, bengene, styrene, and pyridine (all at 20°C), in methylene bromide (at 24°C), in quinoline (at 70°C) and in naphtalene (at 85°C); no dispersion was observed in water, methanol, acetone, toluene, heptane, and cyclohexane, at 20°C. These results show that dispersion takes place in media whose molecules have a four- or six-element "closed ring" structure, or a "double-ring" structure with a melectron configuration: or in media containing a non-saturated radical in the molecule; or in media made up of simple molecules

Card 2/3

S/185/62/007/007/010/010 I048/1248

The hyperacoustic properties of ...

having # -electrons, i.e., in all whose molecules are compact and possess a relatively large number of mobile electrons. The mechanism of the acoustic dispersion in non-dissociated liquids is discussed, and a certain analogy is discovered between the structure of a molecule and its tendency towards fluorescence and acoustic relaxation. There are 2 tables.

ASSOCIATION: Moskovskiy universitet (The University of Moscow)

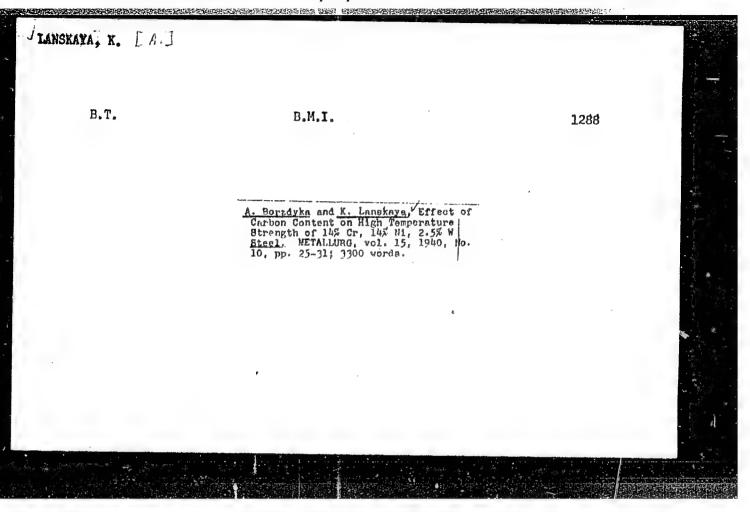
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NOVOZHILOV, V.V., doktor okon. nauk, prof., otv. red.; LANEKAYA,

K.A., red.

[Mathematicoeconomic problems; transactions] Natematikoekonomicheskie problemy; trudy. Leningrad, Lzi-vo Leringr.
univ., 163. 88 p. (MIRA 17:7)

1. Leningradskaya konferentsiya po voprosam primeneniya matematiki v sotsialisticheskoy ekonomike. 1st, 1961.



LANSKAYA, K. A.

Jan/Feb 48

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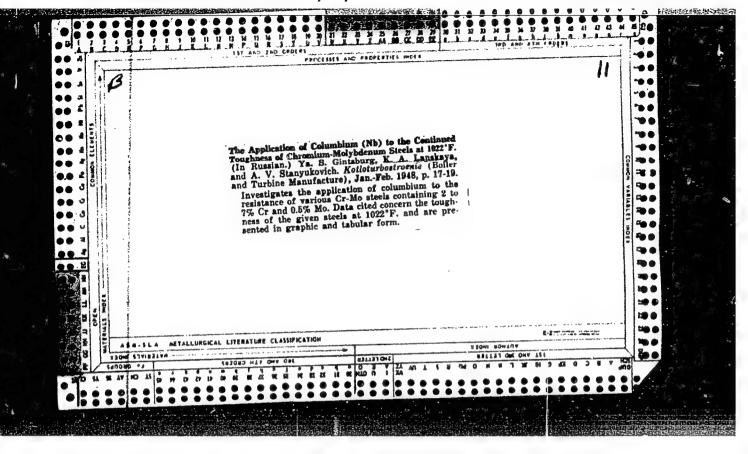
Steel, Chromium Molyrdenum Columbium

"The Effect of Niobium on the Lesting Solidity of Chrome Molybdenum Steel at 550°,"
Ya. S. Gintsburg, Cand Tech Sci; A. V. Stanyukovich; K. A. Lenskaya, Engr, Con Sci
Res Turboboiler Inst imeni I. I. Polsunov, 24 pp

"Kotloturbostroy" No 1

Studies effect of michium on its resistance to prolonged tension of a series of molykeenum chrome steels containing 2+7% Cr and 0.5% No. Gives date on stability of chromemolybdenum-niobium steels at 500°.

PA 1/49T71



ESTULIE, G.V.; ETLHIKOV, A.P.; LABSKATA, K.A.

"Metal testing at elevated temperatures," IA.S. Gintsburg.
Reviewed by G.V. Estulin, A.P. Ryl'nikov, K.A. Lanekaia,
Zav.leb. 21 no.4:509-511 '55 (MLRA 8:6)

1. TSentral'nyy nauchno-iseledovatel'skiy institut chernoy metallurgii.

(Metals -- Testing) (Gintsburg, IA.)

LANSKAYA, K.A.

PHASE I BOOK EXPLOITATION

SOV/2192

18(2)

Pridantsev, Mikhail Vasil'yevich, and Kseniya Alekseyevna Lanskaya

Stali dlya kotlostroyeniya (Steels for the Manufacture of Boilers)

Moscow, Metallurgizdat, 1959. 303 p. 4,500 copies printed.

Ed.: G.K. Shreyber; Ed. of Publishing House: Ye. N. Berlin; Tech. Ed.: P.G. Islent'yeva.

PURPOSE: This book is intended for scientific workers of institutes and educational institutions, and engineers and designers dealing with the production and application of heat-resistant steels.

COVERAGE: The book presents data on changes in the structure and properties of steels subjected to high temperatures and stresses for a long period of time and data on the effect of carbon, alloying elements, impurities, and structural factors on the properties of pearlite and austenite heat-resistant boiler steels. Problems of the theory of creep, heat resistance, and the principles of alloying are discussed. Information is also given on

Card 1/4

30V/2192 Steels for the Manufacture of Boilers the properties of pearlite and austenite heat-resistant steels for boiler installation and on other designs intended for longtime service at temperatures of 500-700° C. The authors thank Senior Scientific Worker R.M. Kireyeva of the Steel Institute of TsNIIChM and laboratory technicians R.A. Raykel'son and L. M. Maksimova. There are 115 references: 74 Soviet, 33 English, 5 German, and 3 French. TABLE OF CONTENTS: Preface 3 5 Introduction Ch. I. Creep and Long-time Strength of Metals and Alloys 9 Ch. II. Structural Instability and Change in Properties of Steels Due to Long-time Service at High Temperatures 27 1. Pearlite spheroidization and coagulation of the carbide 27 phase 32 34 2. Graphitization 3. Aging and formation of new phases Card 2/ 4

Steels for the Manufacture of Boilers	SOV/2192	
4. Thermal brittleness5. Redistribution of alloying elements between solid s and carbide phase6. Diffusion and self-diffusion in iron and steel	43 colution 47 58	
Ch. III. Corrosion Resistance of Steels for Boiler Units 1. Gaseous corrosion 2. Intercrystalline corrosion	67 68 78	-
Ch. TV. Effect of Alloying Elements on the Properities of resistant Steels for Steam Power Installations 1. Effect of carbon 2. Effect of chromium 3. Effect of molybdenum 4. Effect of tungsten 5. Effect of vanadium 6. Effect of titanium 7. Effect of niobium 8. Effect of zirconium	Heat- 82 88 90 92 96 100 113 136	La de la company
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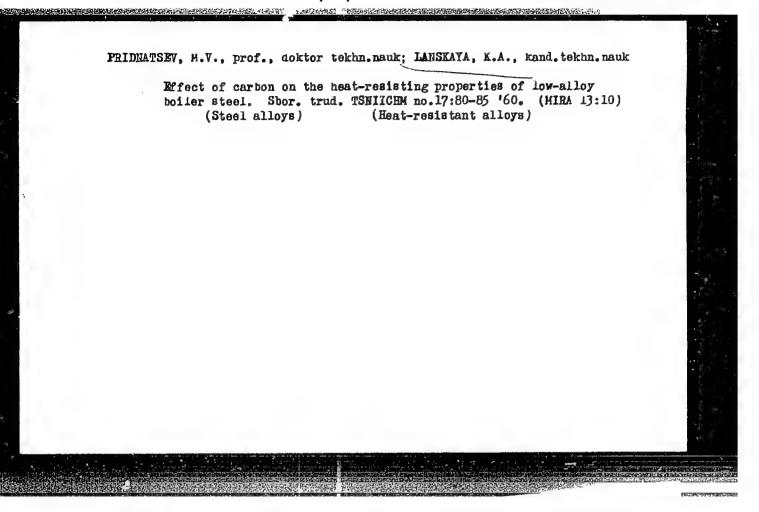
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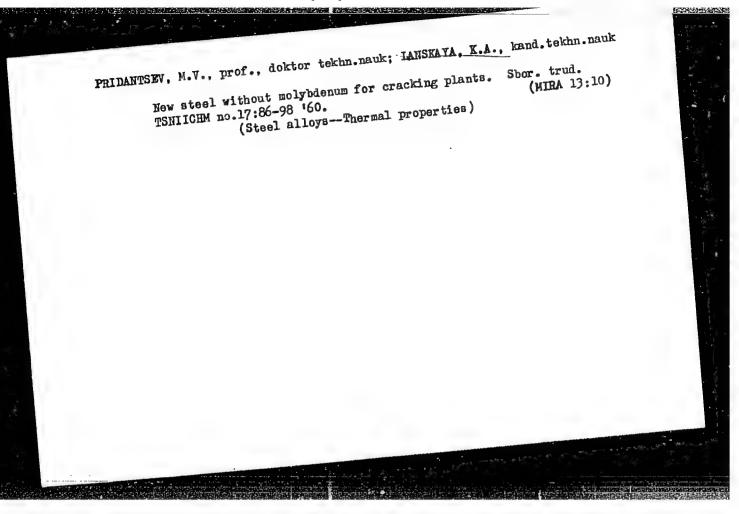
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34530 S/659/61/007/000/017/044 D217/D303

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AUTHORS: Lanskaya. K.A., and Gorchakova, E.N.

TITLE: Microalloying of heat resistant tube steels

SOURCE: Akademiya nauk SSSR. Institut metallurgii. Issledovaniya po zharoprochnym splavam, v. 7, 1961, 169 - 177

TEXT: Small additions of B, Ce, La, Zr, Ca and Ba have found wide application in industry in manufacturing heat resistant and stainless steels and alloys. Many investigations have been carried out within the last few years on the influence of these elements on the properties of various alloys, but the nature and mechanism of this influence are not fully understood. Therefore, the authors made an attempt to discover the mechanism of the influence of each additive both as a deoxidizer and as an alloying element, apart from their influence as modifiers, desulphurizers and elements promoting the formation of high melting point compounds with harmful impurities. For this purpose, the materials were chosen so as to be free of any non-ferrous metals (Pb, Zn, Sn, etc.) and so as to contain a mini-

Card 1/3

S/659/61/007/000/017/044 D217/D303

Microalloying of heat resistant ...

mum of sulphur. New accurate methods for determining small additions were developed for this purpose: Chemical, spectral and spectrochemical analyses. N.N. Sorokina, V.M. Golubeva, F.A. Ozerskaya and A. M. Krichevskaya participated in this work. The investigation was carried out on two steels belonging to different classes, in order to verify the influence of small additions on the properties of α -and γ - base solid solutions of iron. The Cr-Mo-V steel 12XM Φ (12Kh MF) and the Cr-Ni-Nb steel BM694 (VI694) were melted in 10 and 30 kg furnaces. The following additions (in %) were made to these steels: 0.005 - 0.10 B, 0.05 - 0.50 Ca, 0.05 - 0.50 Ba, 0.03 - 1.00 Steels: 0.005 - 0.10 B, 0.05 - 0.50 Ca, 0.05 - 0.50 Ba, 0.03 - 1.00 Steels: 0.005 - 0.10 B, 0.05 - 0.50 Ca, 0.05 - 0.50 Ba, 0.05 - 0.05 Ca, 0.05 - 0.50 Ba, 0.05 - 0.05 Ca, 0.05 - 0.50 Ca, 0.05 - 0.05 Ca, 0 Zr, 0.01 - 0.50 Ce and 0.01 - 0.50 La. All additions were made to the steels after deoxidation with Si, Mn and a nickel-manganese alloy. Cerium was added in the form of mish metal or ferro-cerium, boron as ferro-boron, zirconium as 30 % or 46 % silicozirconium, calcium as silicocalcium, barium as an aluminum-barium alloy and metallic barium; lanthanum was only added to steel 3M694 (EI694). It was found that Ca and Ba act only as deoxidants of steel; they reduce the gas content of the metal and purify it from non-metallic impurities, especially SiO2. Additions of Ce + La and Zr to perli-Card 2/3

Microalloying of heat resistant ... S/659/61/007/000/017/044 D217/D303

tic steel have a deoxidizing effect (basically they reduce the ${\rm Al}_2$ O₂ content of the metal), whereas when added to austenitic steel they also act as alloying elements, strengthening the material. Boron is an active deoxidizer, but its main effect is its ability to act as an alloying element in the grain boundaries of the α - and γ -solid solutions, (which are the weakest portions at high temperatures) owing to the fact that boron is a surface-active element. There are 5 figures, 2 tables and 3 Soviet-bloc references.

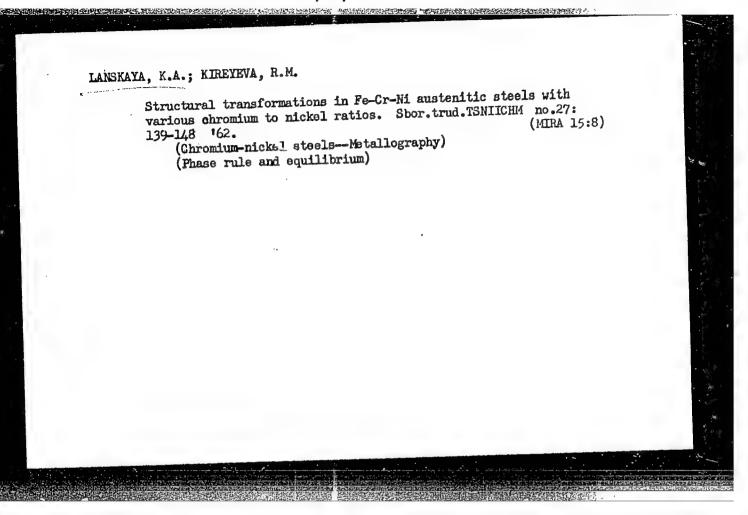
Card 3/3

PRIDANTSEV, M.V., doktor tekhn.nauk, prof.; IANSKAYA, K.A., kand.tekhn.nauk

Development and application of heat resistant pipe steel. Teploenergetika 9 no.8:2-6 Ag '62.

1. TSentral 'nyy nauchno-issledovatel'skiy institut chernoy metallurgii.

(Steel) (Pipe, Steel)



SELECTION CONTROL OF THE SELECTION OF TH

LANSKAYA, K.A., KIRYEVA, R.M.

Structural transformations in Fe-Cr-Ni austenite steels with different Cr/Ni ratio.

SPECIAL STEELS AND ALLOYS (SPETSIAL'NYTE STALI I SPLAYY), Collection of Studies, Issue 27, 240 pages, published by the State Scientific and Technical Publishing House for Ferrous and Non-Ferrous Metallurgy, Moscow, MSSR, 1962.

S/133/63/000/003/004/007 A054/A126

AUTHORS:

Lanskaya, K.A., Kireyeva, R.M., Gorchakova, E.N.

TITIE:

On the quality of 12 X1 M Φ (12KhlMF) grade billets and tubes

PERIODICAL: Stal', no. 3, 1963, 242 - 247

TEXT: Investigations carried out into the mechanical properties of 1.7 12KhlMF grade billets and tubes of various diameter and wall-thickness revealed a considerable non-uniformity as to characteristics, depending on their section, diameter and wall-thickness. In view of the fact that the investigated samples originated from the same grade of steel it could be assumed that this anisotropy in properties must be put down to differences in the heat treatment of billets and tubes. Great deviations were found mainly with respect to notch toughness. The tests on the effect of heat treatment (rate of cooling and annealing temperature) showed that the optimum results as to mechanical properties and heat resistance are obtained upon normalizing at 960 - 980°C and annealing at 730 - 750°C for 3 h (for tubes up to 25 - 30 mm wall-thickness). For thick-walled tubes an increased rate of cooling should be applied by means of pressurized air

Card 1/3

S/133/63/000/003/004/007 A054/A126

On the quality of 12 X1 MT (12Kh1MF)

or water-oil cooling after heating to 960 - 980°C with subsequent annealing. The respective tests were carried out at the TsNIIChM applying 15 different cooling rates. Over-heating and under-heating had varying effects on the properties. Annealing at 800 - 830°C ensures a notch tourness of 20 - 25 kgm/cm2 but deteriorates heat resistance. The anisotropy in mechanical characteristics can be reduced by ensuring that in the heat treating furnaces the tubes are heated uniformly lengthwise and across, moreover, by applying devices which increase the cooling rate. Uniform values for notch toughness, for instance, were obtained at a cooling rate of 36°C/min. There is also a difference in mechanical properties for transverse and longitudinal samples. Low values can be found for transverse contraction and extension of transverse samples cut out from billets, whereas this is not observed in longitudinal specimens. This is explained by the higher gas content (mainly hydrogen), a higher amount of nonmetallic inclusions and a higher degree of deformability of some heats. In general, no direct relationship could be established between the properties of the billet and those of the finished tube. With the present method of assessing the quality, carried out for billets (over 140 mm in diameter) on longitudinal specimens cut out from 90 mm squares and on transverse specimens cut from the finished tube,

Card 2/3

On the quality of 12 X1 M Φ (12Kh1MF)

S/133/63/000/003/004/007 A054/A126

their characteristics cannot be compared. To render this possible, i.e., to make the properties of billets and tubes comparable, both should be investigated by reference to transverse specimens. The investigations and tests described refer to the Yuzhnotrubnyy zavod (Yuzhnotrubnyy Plant) and the Cheiyabinskiy truboprokatnyy zavod (Chelyabinsk Tube-Rilling Plant). There are 7 figures.

ASSOCIATION: LHMNYM (Tanlichm)

Card 3/3

PRIDANTSEV, M.V., doktor tekhn.nauk, prof.; LANSKATA, K.A., kand.tekhn.nauk

Safety factor and choice of permissible stresses in the calculation of boiler pipes. Teploenergetika 10 no.1:61-64 Ja '63.

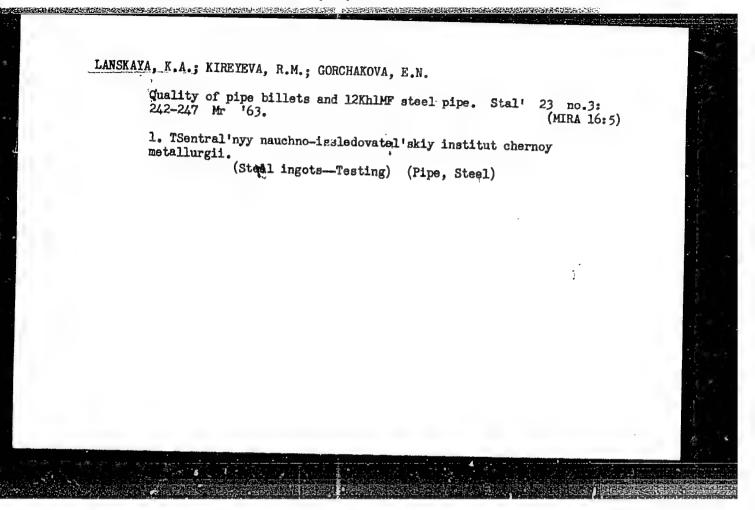
(MRA 16:1)

1. TSentral'nyy nauchno-issledovatel'skiy institut chernoy motallurgii.

(Boilers) (Steampipes)

L 12896-63 EMP(q)/EWT(m)/BDS AFFTC/ASD ACCESSION NR: JD AP3000676 s/0096/63/000/006/0002/900E AUTHOR: Lanskaya, K. A. (Candidate of technical sciences) TITLE: Structure and properties of steels used in boilers at high and ultra-high SOURCE: Teploenergetika, no. 6, 1963, 2-6 TOPIC TAGS: steel pipe, microstructure, heat resistance, heat treatment, durability ABSTRACT: Experiments were conducted to find out the influence exerted by the structure of steel on its heat resistance, to study the stability of hardened and tempered steel structures subjected to continuous heating, and to determine the heat resistance of alloy steels with metastable structures. Because of increasing temperatures and pressures of the steam used in boiler and turbine plants and because of the change to machinery demanding higher heat resistance, these problems have become important in recent years. The influence of structure on the sustained strength of steel pipes with various compositions and mechanical properties was tested and the results tabulated. It was concluded that certain heat treatments for pipes must be selected from the results of sustained strength tests, not from mechanical characteristics determined in tensile and impact experiments. Mechanical

properties (especially resistance to impact) may serve as indicators of the heat treatment to which the steel was subjected, but do not project the sustained resistance to heat of these steels. Moreover, similar mechanical properties may be secure the desired heat resistance. Considering the properly chosen type will structure of steel and its heat resisting qualities, the author recommends microtructural control of pipe steel. Orig. art. has: 7 figures and 1 table. SSOCIATION: TSNIICHERMET	L 12896-63 ACCESSION NR: AP3000676	The second secon				
JEMITTED: 00 DATE ACQ: 21Jun43 ENCL: 00 JE CODE: 000 OTHER: 000	properties (especially restreatment to which the steresistance to heat of these produced by various types secure the desired heat restructure of steel and its structural control of pipe	of heat treatment, is istance. Consider heat resisting qual steel. Orig. art.	similar mechabut only the print the maleri	anical proper roperly chose	ined ties may be n type will	
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ACCESSION NR: AP4012428

\$/0129/64/000/002/0013/0018

AUTHORS: Lanskaya, K.A.; Gorchakova, E.N.; Kireyeva, R.M.

TITLE: Structural transformation in 12KhlMF steel during heat

SOURCE: Metalloved. i term. obrab. metallov, no. 2, 1964, 13-18

TOPIC TAGS: structural transformation, 12KhlMF steel, heat treatment, chrome molybdenum vanadium steel, impact strength, vanadium carbide, yield strength, yield point, hardness

ABSTRACT: Due to high heat resisting properties, chromiummolybdenum-vanadium steel forced chromium-molybdenum steel out of
the reactor production. It was established that vanadium in such
steel strengthens the solid solution and decreases the rate of diffusion processes of elemental redistribution, particularly the
molybdenum. In addition, the presence of thermally-stable, finelydispersed vanadium carbides inhibits the development of displace-

Card 1/3

ACCESSION NR: AP4012428

ment processes during plastic deformation. However, low values of impact strength are observed at room temperature in many chromium—molybdenum-vanadium steel products. To establish the reason for this, the structure and properties of chromium-molybdenum-vanadium leats, the structure and properties of chromium-molybdenum-vanadium leats melted at the "Krasny*y Oktyabr" factory in 140 ton open hearth furnaces. During continuous cooling of 12KhlMF steel, the transformation of austenite can proceed in 3 zones depending on the cooling rate: ferrite-perlite, interstitial and martensite. Components of different sizes are then cooled at one rate by changing cooling conditions. Tempering of hardened or normalized 12KhlMF steel at 600-650C causes separation of finely dispersed vanadium carbides and accompanied by an increase of the yield strength, yield point, and hardness and a decrease of impact strength. With an increase in tempering temperature, agglomeration of vanadium carbides occurs which decreases strength properties and increases plastic properties and impact toughness. During tempering of annealed steel, vanadium carbides are not separated and mechanical properties

Card 2/3

ACCESSION NR: AP4012428

vanadium carbides were fully separated in the cooling process during annealing. Low and unequal values of impact strength in heat-treated, thick-walled tubes were observed due to an insufficient cooling rate and break in temperature during tempering in factory furnaces. High heat resisting properties with sufficiently high temporary mechanical properties were reached after heating at 960-980C, cooling from this temperature at a rate of no less than 200-300 degrees/min., and tempering at 730-750C. Orig. art. has: 4 figs., 3 tables.

ASSOCIATION: TSNIICHM

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DATE ACQ: 03Mar64

ENCL: 00

SUB CODE: ML

NO REF SOV: 003

OTHER: 000

Card 3/3

Management of the property of

LANSKAYA, K.A., kand. tekhm. nauk

Froperties of thick-walled pipes from 12KhlMF steel, their bends, and welded joints. Teploenergetika 11 no.12*9-14 D '64 (MIRA 18*2)

1. TSentral'nyy nauchno-issledovatel'skiy institut chernoy metallurgii imeni I.P.Bardina.

L 14959-65 EVT (m)/ENA(d)/EWP(t)/EWP(b) AFWL/ASD(m)-3 MJW/JD/MLK

ACCESSION NR: AT4046856

S/0000/64/000/000/0284/0290

AUTHOR: Lanskaya, K. A.

TITLE: Structure as a factor in the heat-resistance of boller steel

SOURCE: AN SSSR. Nauchny*y sovet po probleme zharoprochny*kh splavov.
Issledovaniya staley i splavov (Studies on steels and alloys). Moscow, Izd-vo Nauka, 1964,
284-290.

TOPIC TAGS: boiler steel, perlite steel, steel structure, cooling rate, austenitic transformation, steel heat resistance, steel mechanical property/steel 12Kh1MF, steel

ABSTRACT: The author discusses the effect of cooling rate (1 to 1000 degrees centigrade/min.) on the structure, stress-rupture strength and mechanical properties of two steels of the perlite type: 12Kh1MF with 1% Cr, 0.25-0.35% Mo and 0.15-0.30% V, and 12Kh2MFSR with 1.5-1.8% Cr, 0.5-0.8% Mo, 0.15-0.30% V, 0.005% B and an unspecified 12Kh2MFSR with 1.5-1.6% Cr, 0.5-0.8% Mo, 0.15-0.30% V, 0.005% B and an unspecified percentage of Si. On the basis of thermokinetic diagrams of the steels, revealing the course of austenite transformations in the ferrite-perlite region (at 1-6 degrees/min.), intermediate region (at 200-250 degrees/min.) and martensite region (at up to 1000 degrees/min.) as a function of cooling rate, samples with ferrite-perlite, intermediate

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ACCESSION NR: AT4046856

and nartensite structures, desired for tests, were prepared by appropriate thermal treatment. From the results, graphs of stress-rupture strength were plotted in a logar-ithmic system of double strain vs time allowing an extrapolation to 10,000 and 100,000 hrs. A highly tempered martensite structure was found to be the most effective positive factor for heat-resistance, considerably exceeding the intermediate and ferrite-perlite structures in promoting stress-rupture strength. "E. N. Gorchakova took part in the preparation of the thermokinetic diagrams." Orig. art. has: 4 figures and 1 table.

ASSOCIATION: None

SUBMITTED: 16Jun64 ENCL: 00 SUB CODE: MM

NO REF SOV: 001 OTHER: 002

Cord 2/2

CC NR: AP5026533	SOURCE CODE: UR/0286/65/000/019/0073/0073	
VENTOR: Lanskava, K. A.: Goro	chakova, E. N.; Surovtseva, Ye. D.; Lapitskaya, Ye. M.;	
alysheva, V. P.; Zemzin, V. N		
Y4,55. 44,	55	
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ssledovatel'skiy institut che	ernoy metallurgii)]	
	The same of the sa	
OURCE: Byulleten' izobreteni	y i tovarnykh znakov, no. 19, 1965, 73	
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	bium containing steel, tungsten containing steel	-
N	27	
	cate introduces a ferritic steel containing silicon, m, vanadium, niobium, and tungsten. In order to in-	•
	trength, the steel has the following composition in %:	
	-1.0 Mn, 2.0-10.0 Cz, 0.5-2.0 Mo, 0.15-0.50 V,	
5-1.5 Nb, and 6-10 W.	[ww]	
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ANTIKE: Lanskaya, K, A. (Candidate of	B
TITLE: Grade E1695R Cr-Ni-W-No steel	ith boron
7, 1965,	41-46
heat resistant steel, aus	tenitic steel, boiler steel, Laves phase, X superheater tube, steam line tube/ KI695R
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Steel (IKh14010ASOK Breez)	(LwsQuopp) ere pre-
ABSTRACT: The results of an investige	resistant austenitic boiler steel. The steel resistant austenitic boiler steel. The steel resistant properties for 700°C. The high heat-resistant properties
Bented. This same at 6	resistant austenitic boller state properties 50-700°C. The high heat-resistant properties with tungsten miobium; boron; and cerium.
is designed for longuing at a silvering	WILLIAM WARREN THE THE PART OF
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period of time is assured by keeping the Cr/Ni ratio at less than unity. The chemical composition of EI695R steel is as follows: 0.07-0.12% C, < 0.60% Si, 1.0-2.0% Nm, 13.0-15.0% Cr, 18.0-20.0% Ni, 2.0-2.75% W, 0.9-1.3% Nb, < 0.30% Cu, < 0.020% S, < 0.030% P, 0.002-0.005% B, 0.02% Ce. Phase chemical analysis and Xray micrography of specimens of E1695R steel austenitized at 1100-1150°C revealed the presence of carbonitride phase of nichium Nb(CN). The kinetics of the transformations that occur in this steel in the course of its longtime aging (at 650-900°C for J000-5000 hr) is as follows: γ-solid solution + Nb(CN) → γ-solid solution + Laves phase + Nb(CR). The Laves phase, though the amount in which it is present is very small (not more than 1%), may markedly affect the steel's resist-size to high temperatures. During aging under stress, which is analogous to longtime tensile tests or performance at high temperatures and pressures, the Leves phase forms mostly in the grain body, along the slip lines, thus strengthening the grain body and so increasing heat resistance. The plastic properties of the steel following its longtime strength tests are sufficiently high, and so are its mechanical properties. Thus, for example, after 17,000 hours of cresp tests and tensile tests, no substantial change occurred in the mechanical properties of the speci-The physical properties of steel EI695R include: specific weight 8.1 g/cm3; heat combuctivity \ = 0.036 cal/cm sec °C. All this warrants recommending steel

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그렇게 하는 하고, 나는 집에 가를 하는 것을 하고 있다면 하면 하는 것이 없는 것이 없는 것이다.				
E1695B for broader use in well as of fittings opera	ting in installa	of superher	ater and steam-111 superhigh steam p	ne tubes as arameters.
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